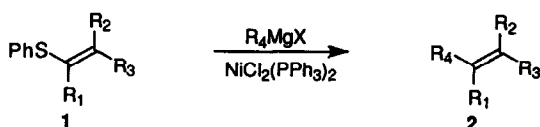


A STEREOSELECTIVE ROUTE TO *E*-3-HYDROXYALKENYL PHENYL SULFIDES

Seiichi Takano,* Yoshiaki Sugihara, and Kunio Ogasawara
 Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan

Summary: *E*-3-Hydroxyalkenyl phenyl sulfides are generated stereoselectively when 2,3-epoxyalkyl phenyl sulfides having 3-*syn*-substituent are exposed to an excess *n*-butyllithium in THF solution.

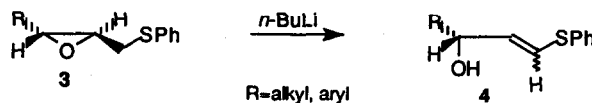
Phenyl sulfide on alkenyl sp² carbon is a good leaving group in the cross-coupling reaction with Grignard reagents in the presence of a nickel catalyst¹⁾ (Scheme 1). Since the reaction proceeds with retention of the olefin configuration, stereoselective preparation of the alkenyl phenyl sulfide precursors is the most



Scheme 1

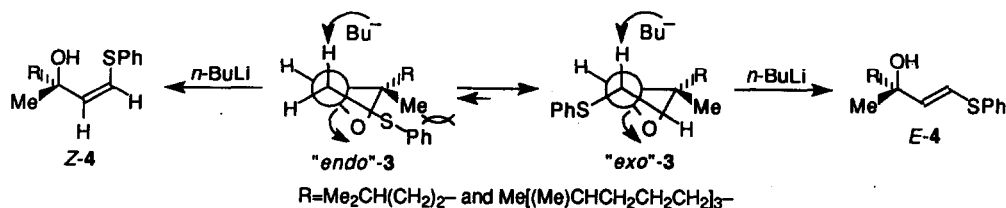
important. Herein we report a stereoselective method for the preparation of *E*-3-hydroxyalkenyl phenyl sulfides starting with 2,3-epoxyalkenyl phenyl sulfides having 3-*syn*-substituent.

Quite recently we found²⁾ an efficient formation of a *E/Z*-mixture of 3-hydroxyalkenyl phenyl sulfides (4) by treating 2,3-epoxyalkyl phenyl sulfides (3) with an excess *n*-butyllithium in THF (Scheme 2). In this

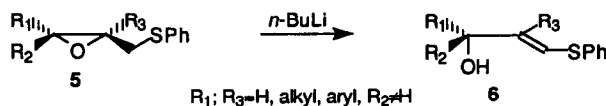


Scheme 2

transformation we observed that the trisubstituted epoxides generated from geraniol and phytol showed an exceptionally high stereoselectivity to furnish the corresponding alkenyl phenyl sulfides (4) having *E* configuration. Since we reasoned that this striking difference is due to the steric repulsion between the 3-*syn*-methyl group and the thiophenyl group forcing the substrates to take "*exo*" transition state (*exo*-3) to give rise to the *E* olefin products (*E*-4) (Scheme 3), we now choose the epoxy sulfides (5) having 3-*syn*-substituent as the substrates so as to obtain the alkenyl phenyl sulfides (6) having *E* configuration in a selective manner.



Scheme 3



Scheme 4

Table 1 Reaction of the Epoxy Sulfides (5a-j) with *n*-Butyllithium

Entry	epoxy sulfide ¹ (5)		alkenyl sulfide (6)	yield ² (%)	ratio ³ (<i>E</i> : <i>Z</i>)
	R ₁ :R ₂ :R ₃				
1	5a	H:BnOCH ₂ :H	6a	91.9	8.8 : 1
2	5b	H: <i>n</i> -C ₅ H ₁₁ :H	6b	83.3	10.8 : 1
3	5c	H:BnOCH ₂ CH(OH)CH ₂ :H ⁴	6c	99.4	13.2 : 1
4	5d	H:BnOCH ₂ CH(OH)CH ₂ :H ⁵	6d	84.9	6.3 : 1
5	5e	Me:Me:H	6e	75.4 ⁶	>99 : 1
6	5f	-(CH ₂) ₅ -:H	6f	68.8 ⁷	>99 : 1
7	5g	Me:Me ₂ CH(CH ₂) ₂ :H	6g	87.2	>99 : 1
8	5h	-(CH ₂) ₅ -:Me	6h	86.2	8.7 : 1
9	5i	Me ₂ CH(CH ₂) ₂ :Me:H ⁸	6i	93.9	31 : 1
10	5j	Me[(Me)CHCH ₂ CH ₂ CH ₂] ₃ :Me:H ⁸	6j	88.9	>99 : 1

1. prepared by treating the alcohol precursor with diphenyl disulfide in the presence of tri-*n*-butylphosphine.³
 2. isolated yield. 3. determined by ¹H-Nmr (500 MHz). 4. (2*R*,3*S*,5*S*). 5. (2*S*,3*R*,5*S*). 6. 15.2% of the starting material was recovered. 7. 9.1% of the starting material was recovered. 8. Ref. 2.

Three types of the 2,3-epoxyalkyl phenyl sulfides, 3-*syn*-disubstituted (**5a-d**), 3-*syn*-3-*anti*-trisubstituted (**5e-g**), 3-*syn*-3-*anti*-tetrasubstituted (**5h**), were subjected to the reaction conditions. Thus, treatment of **5** with 1.5 equiv. of *n*-butyllithium in THF at -78 °C for 30 min afforded the sulfides **6** in excellent yields (Table 1). Among these the trisubstituted substrates (**5e-g**) furnished 3-hydroxyalkenyl phenyl sulfides (**6e-g**) having *E* configuration exclusively in an excellent yield (entries 5-7). Both the disubstituted and the tetrasubstituted epoxides, on the other hand, furnished the corresponding *E*-olefins (**6**) in excellent yields in ratios of 6.3-13.2:1 though they did not exhibit complete stereoselectivity (entries 1-4 and 8) (Table 1). It seems that the stereoselectivity is mostly due to the 3-*syn*-substituent on the epoxides, but the secondary effect by the remote hydroxy group seems to be also significant (compare entries 3 and 4). Since the present procedure may be coupled with the Katsuki-Sharpless epoxidation reaction, it is particularly useful for the preparation of optically active *E*-3-hydroxyalkenyl phenyl sulfides.

References and Notes

- (1) H. Okamura, M. Miura, and H. Takei, *Tetrahedron Lett.*, **1979**, 43; H. Okamura and H. Takei, *Tetrahedron Lett.*, **1979**, 3425; E. Wenkert, T. W. Ferreira, and E. L. Michelotti, *J. Chem. Soc., Chem. Commun.*, **1979**, 637; H. Takei, H. Sugimura, M. Miura, and H. Okamura, *Chem. Lett.*, **1980**, 1209; H. Sugimura, H. Okamura, M. Miura, M. Yoshida, and H. Takei, *Nippon Kagaku Kaishi*, **1985**, 416; H. Sugimura and H. Takei, *Bull. Chem. Soc. Jpn.*, **1985**, **58**, 664.
- (2) S. Takano, Y. Sugihara, and K. Ogasawara, *Synlett*, **1992**, 668.
- (3) I. Nakagawa and T. Hata, *Tetrahedron Lett.*, **1975**, 1409; I. Nakagawa, K. Aki, and T. Hata, *J. Chem. Soc., Perkin Trans. 1*, **1983**, 1315.

(Received in Japan 17 September 1992)